

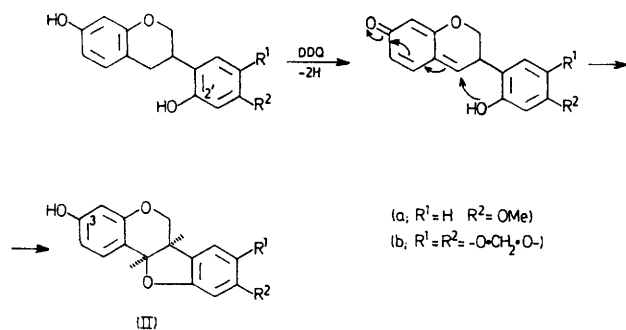
A Possible Chemical Analogy for Pterocarpan Biosynthesis

By MARA CORNIA and LUCIO MERLINI*

(Istituto di Chimica, Centro del C.N.R. per le Sostanze Organiche Naturali, Politecnico di Milano, 20133 Milano, Italy)

Summary Oxidative conversion of 2'-hydroxyisoflavans to pterocarpan by 2,3-dichloro-5,6-dicyanobenzoquinone is suggested as a possible chemical analogy for the corresponding biosynthetic pathway.

ALTHOUGH the biogenesis of flavonoids has been well studied¹ direct experimental evidence for the interconversion *in vivo* of different classes of isoflavonoids is available only in a few cases.² In particular, there is no evidence for the interconversion of pterocarpan and isoflavans. On the



basis of the characteristic 2'-hydroxylation pattern of natural isoflavans and their chemical analogues, it could be suggested that isoflavans are derived by reductive ring-opening of pterocarpan.² On the other hand, Grisebach and his co-workers³ have shown that 2'-hydroxylation of an isoflavone or isoflavone is a step in the biosynthesis of coumestans.

We now report a new chemical correlation that could suggest that the conversion of 2'-hydroxyisoflavans to pterocarpan might also occur *in vivo*. Mild oxidation of the isoflavans (Ia) or (Ib) with 2,3-dichloro-5,6-dicyanobenzoquinone (DDQ) in benzene at room temperature for 8 h afforded demethylhomopterocarpan (IIa) or maackiain (IIb) in 30% yield (n.m.r., direct t.l.c. comparison). The reaction most probably proceeds through a quinonemethide intermediate, which undergoes a nucleophilic addition by the 2'-OH group. Similar behaviour shown by 2,4'-dihydroxydiphenylethane⁴ and 2,2'-dihydroxydiphenylethane⁵ supports this hypothesis. The 3-OH (or 3-OR) substitution in all the natural pterocarpan so far found in Nature is consistent with the suggested biosynthetic pathway, which requires such an activating group.

Shortage of starting materials (prepared by hydrogenation of the corresponding (II), extracted from *Swartzia madagascariensis* heartwood⁶) prevented a closer investigation of possible minor products of the reaction. However, the ready further oxidation of (II) to coumestans has been already demonstrated by Grisebach,⁷ and Roux⁸ and their co-workers.

We thank Mr. D. B. Fanshawe, Director of Forest Research, Kitwe, Zambia, for providing the *Swartzia* heartwood.

(Received, 17th March 1975; Com. 316.)

¹ H. Grisebach, in 'Recent Advances in Phytochemistry,' ed. T. J. Mabry, North Holland, Amsterdam, 1968, p. 379.

² E. Wong, *Fortschr. Chem. org. Naturstoffe*, 1970, **28**, 1.

³ J. Berlin, P. M. Dewick, and H. Grisebach, *Phytochemistry*, 1972, **11**, 1689.

⁴ K. Schofield, R. S. Ward, and A. M. Choudhury, *J. Chem. Soc. (C)*, 1971, 2834.

⁵ B. Cardillo, M. Cornia, and L. Merlini, *Gazzetta*, submitted for publication.

⁶ S. H. Harper, A. D. Kemp, W. G. E. Underwood, and R. V. M. Campbell, *J. Chem. Soc. (C)*, 1969, 1109.

⁷ P. M. Dewick, W. Barz, and H. Grisebach, *Chem. Comm.*, 1969, 466.

⁸ D. Ferreira, C. v. d. M. Brink, and D. G. Roux, *Phytochemistry*, 1971, **10**, 1141.